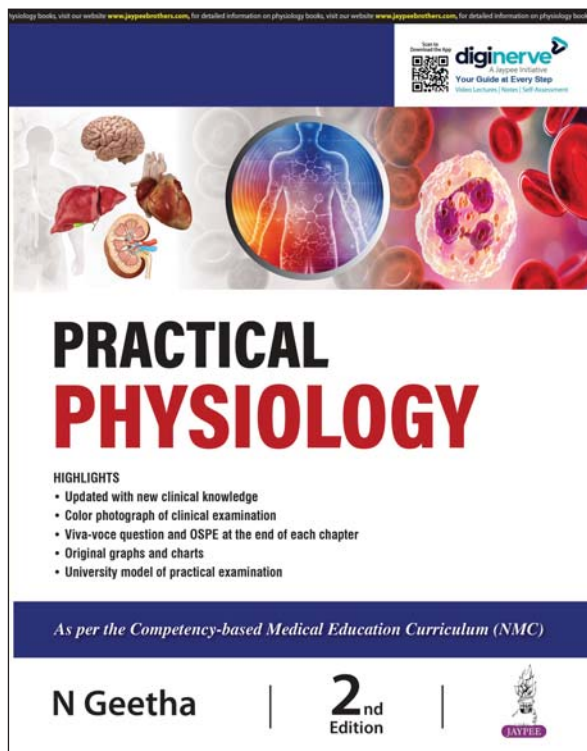


PRACTICAL PHYSIOLOGY

*As per the Competency-Based Medical
Education Curriculum (NMC)*

2nd
Edition

N Geetha



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Why to Buy this Book ?

- The book is clinically oriented, laying emphasis on the physiological basis of the observation.
- The full colored photographs and illustrations help to precisely understand the clinical condition.
- Clinical examination follows the basic pattern, which includes a thorough and systematic approach in examining the patients.
- The book also covers practical aspects on animal experiments and provides graphs and records of the same.
- Viva-voce questions with answers are included at the end of each experiment.
- Scheme of practical examination with sample questions is provided in the last chapter.
- Detailed index is provided at the end of the book.
- This book is useful not only for the medical students, but also for dental, paramedical, and allied health sciences students.

Learning objectives introduced at the start provide a focused overview of the chapter's main themes.

6
Chapter

LARNING OBJECTIVES

- List the different methods of estimation of hemoglobin
- Estimate hemoglobin content of your blood by Sahl's acid hematin method
- Advantages and disadvantages of Sahl's method
- Principle of Sahl's method
- Mention the normal values of hemoglobin concentration in infants, adult and old age
- Why is hemoglobin content more in males
- Name the conditions where there is variation in the hemoglobin content of blood
- Define anemia and mention the types of anemia depending on hemoglobin content
- Describe the steps in the synthesis of hemoglobin
- List the derivatives of hemoglobin
- Describe different types of anemia

PY2.11 Estimate hemoglobin, red blood cell count, total leukocyte count, RBC indices, differential leukocyte count, blood groups, BT/CT.

COLOIMETRIC ESTIMATION OF HEMOGLOBIN

- Different color of
- Sahl's method
 - Talquist's method
 - Haldane's method
 - Cyanmethemoglobin
 - Gaometric

Principle

The hemoglobin present in a measured sample of blood is converted into a derivative having a definite tinge of color. In Sahl's method acid hematin is prepared. In Talquist's method the color of oxyhemoglobin is compared. In Haldane's method, carboxyhemoglobin is prepared. Cyanmethemoglobin derivative is also used to give accurate results. The density of this color is compared with a standard solution of the same derivative. Iron estimation or spectrophotometry is used while deciding the strength and the color of the standard.

Sahl's Method

Principle

Sahl's method is the commonly employed method for hemoglobin estimation. The hemoglobin present in a particular volume of blood is converted to acid hematin by treating it with N/10 HCl. Acid hematin is brown in color and the solution is diluted with distilled water till the color matches with the standard color. The standard color is prepared by treating a sample of blood containing 14.5 g

PY2.11: Estimate hemoglobin, red blood cell count, total leukocyte count, RBC indices, differential leukocyte count, blood groups, BT/CT.

Initial competencies outline essential skills and outcomes.

Full-color photographs and illustrations enhance precise understanding of the clinical condition.



Fig. 6.3: Correct matching of sample with standard color.



Fig. 6.4: Overdilution of blood sample, color not matching with the standard color.



Fig. 6.1: Sahl's hemoglobinometer showing the box with standard color, brush, diluting tube, pipette, glass rod, N/10 HCl and dropper.



Fig. 6.2: Sahl's hemoglobinometer showing the color standards.

- of the glass block. While comparing, fill the upper above the level of the solution but it should not be completely taken out of the tube. If comparison is done with the glass rod in the solution, the solution will appear lighter. While comparing the color, avoid viewing through the markings on the tube and hold the apparatus against natural light.
- Add water drop by drop, mix and compare each time after every addition.
 - Continue till the color matches (Fig. 6.3)
 - Rotate the glass rod and take the reading in g/dl.
 - Read the lower meniscus. Since the solution is transparent, the lower meniscus is clearly seen even though it is colored.

Precautions

- The color of the standard should be examined from time to time to check any fading or change in color.
- After pricking, the finger should not be squeezed to obtain blood. This may give a false low value for hemoglobin since the blood gets diluted with tissue fluid. A few flowing drops should be obtained.
- There should not be any air bubbles in the blood column in the pipette.
- Blood from the pipette should be immediately transferred into the tube containing N/10 HCl. Otherwise blood may clot.

Result

Read directly from the diluting tube and record the result in grams of hemoglobin per 100 ml. of blood.

Disadvantages of Sahl's Method

- There can be visual error.
- This method estimates only oxyhemoglobin and reduced hemoglobin. Carboxy, met and sulfhemoglobin cannot be converted to acid hematin.
- The color of the standard fades over time.
- If the reading is not taken immediately, the color of acid hematin changes.

Chapter 8: Estimation of Hemoglobin (Hemoglobinometer)

of hemoglobin per 100 mL of blood with N/10 HCl and diluting it 100 times. So 100% reading in the diluting tube corresponds to 14.5 g/L of hemoglobin.

Apparatus

Sahlb's hemoglobinometer set (Fig. 6.3) contains the following:

- A rectangular plastic box with two color standards on either side with a central provision for keeping the diluting tube. The color standard is made of non-fading, standardised, golden brown glass rods (Fig. 6.2).
- Specially graduated diluting tube in square or round shape. It is graduated in percentage (20–100%) on one side and in g/dL (12–24 g/dL) on the opposite side.
- A glass rod with flat tip to stir the contents.
- Hemoglobin pipette with a 20 mm³ mark and rubber tubing with a metal piece to suck the blood. There is no bulb in the hemoglobin pipette and the 20 mm³ mark indicates a definite measured volume and not an arbitrary volume as in the case of BCC and WBC pipette.
- Needle or lancet.
- Bottle containing distilled water and a dropper with it.
- Bottle containing N/10 hydrochloric acid, which is prepared by mixing 1 mL of concentrated HCl and 99 mL of distilled water.
- Break to form the tube.



Fig. 6.3: Sahlb's hemoglobinometer showing the box with standard color, diluting tube, pipette, glass rod, N/10 HCl and dropper



Fig. 6.2: Sahlb's hemoglobinometer showing the color standards

Procedure

- See that the hemoglobin pipette is clean and dry.
- Place the graduated tube between the standards in the plastic box.
- Fill it with N/10 HCl up to the lowest mark.
- Prick the tip of the finger to get a large drop of blood.
- Hold the pipette in the right hand with the graduation in front.
- Place the mouthpiece between the lips.
- Hold the pipette at an angle of 45° above the horizontal and touch the tip of the pipette to one side of the drop.
- Suck the blood till the blood reaches the 20 mm³ mark (if any air bubble enters the pipette, repeat with a new dry one).
- Remove the pipette and clean the outer surface with a cotton swab by wiping it towards the tip to remove any blood that is present on the outer surface of the pipette. Do not touch the open end while wiping because blood will come out of the pipette due to capillary action.
- If the blood has slightly crossed the mark, take the excess blood out by touching the tip of the pipette on the palm of the hand for a couple of times and the blood will recede back in the pipette. Make sure that you are wearing gloves if blood is taken from another person.
- Do not use a swab or a filter paper to remove the excess blood. It will absorb a very large quantity of blood.
- While transferring the blood to the HCl in the diluting tube, blow down slowly.
- Dip its tip in the acid and blow out gently to transfer all the blood into the tube.
- Suck the superficial acid and rinse the pipette repeatedly till all the blood in the pipette is transferred into the diluting tube.
- Mix the blood and the acid with the glass rod provided.
- Note the time. Wait for 10 minutes to allow the brown color of acid hematin to develop. The color does not develop immediately. Its intensity changes with time. 95% of the color is reached by the end of 10 minutes.

Cyanmethemoglobin Method

Principle

The basis of this method is dilution of blood in a diluent called Drabkin cyanide-ferricyanide solution. Ferrous ion of hemoglobin is converted to ferric ion by ferricyanide and methemoglobin so produced is combined with potassium ferricyanide to produce cyanmethemoglobin which is red colored. All hemoglobin derivatives of blood except sulfhemoglobin are measured by this technique. But sulfhemoglobin is rarely present in significant amounts in blood. *Cyanmethemoglobin method is one of the most reliable and the most accurate method and is the WHO'S recommended method for determining the hemoglobin concentration of blood.* The color of cyanmethemoglobin in the unknown blood is compared with the standard solution of cyanmethemoglobin.

The book thoroughly covers principles and procedures, ensuring each topic is easy for students to understand.

Viva-voce questions with answers and a structured, objective-based practical exam are provided at the end of every chapter.

OBJECTIVE STRUCTURED PRACTICAL EXAMINATION

I. Suck blood in the hemoglobin pipette for estimating your hemoglobin concentration

1. Take a clean and dry hemoglobin pipette
2. Sterilize the finger tip
3. Give a deep prick using a sterile lancet (do not squeeze the finger tip for obtaining blood)
4. Wipe off the first drop using sterile cotton
5. Apply gentle pressure so that a drop of blood appears on the finger tip
6. Suck blood into the pipette exactly up to the 20 mm³ mark on the pipette
7. Wipe the tip of the pipette
8. Press the finger tip with clean cotton

II. Do the hemoglobin estimation with the blood taken in the hemoglobin pipette

1. Take a clean diluting tube
2. Take N/10 HCl up to the lower mark in the tube
3. Blow the blood in the pipette into the acid solution in the diluting tube.
4. Rinse the pipette in the acid solution 2 more times and blow out into the diluting tube so that whole of blood remaining in the pipette is transferred to the diluting tube
5. Place the diluting tube between the standard color
6. Note the time and wait for 10 minutes for the formation of acid hematin
7. Add distilled water drop by drop mixing the solution with the glass rod after the addition of each drop.
8. After adding two or three drops of water and mixing, compare the color of the solution with the standard color of the glass block. While comparing, lift the strainer above the level of the solution but it should not be completely taken out of the tube.
9. Add water drop by drop, mix and compare each time after every addition.
10. Continue till the colors match.
11. Raise the glass rod and take the reading in g/dL.
12. Read the lower meniscus.

VIVA QUESTIONS

1. Give the normal hemoglobin content.
 - At birth: 20–25 g/L
 - Adult male: 14–16 g/L
 - Adult female: 12–15 g/L
2. Mention the variations in hemoglobin content.
 - Increase in hemoglobin content is seen in newborns, at high altitude, after exercise, haemorrhage, polycythemia vera, etc.
 - Technical error like increase in the amount of blood taken in the pipette (20 µL) gives a higher value.
 - Decrease in hemoglobin content is seen in anemia. All conditions producing anemia will result in decrease in hemoglobin content.
 - In hemodialysis as in pregnancy, Hb concentration will be less than normal.
 - Technical error like less amount of blood taken in the pipette and blood sample obtained by squeezing the fingertip will give a lower value. When the fingertip is squeezed, the fluid coming out contains more of tissue fluid and less number of RBC. Failure to wait for 10 minutes after mixing blood with N/10 HCl will give a low hemoglobin value. If the amount of N/10 HCl taken is not adequate, all the hemoglobin present in the sample will not be converted to acid hematin and a low value will be obtained.
3. What happens if hemoglobin is found in plasma as the free hemoglobin?
 - If Hb is present dissolved in plasma and not confined within the RBC, it leads to the following problems:
 - It increases the viscosity of blood leading to an increase in the blood pressure.
 - Increase in the osmotic pressure of blood.
 - Free hemoglobin will be filtered and excreted by the kidneys leading to hemoglobinuria and precipitation of hemoglobin in the renal tubules leading to renal failure and anemia.
 - Free hemoglobin will be rapidly destroyed by the reticuloendothelial cells leading to increase in the bilirubin content of blood.
4. What is the reason for increase in hemoglobin content in newborns?
 - During fetal life the type of hemoglobin in blood is fetal (fetal hemoglobin). It contains two α and two γ chains. The γ chain does not combine with 2,3-bisphosphoglycerate (2,3-BPG) and so HbF has greater affinity for oxygen. It releases less O₂ at the tissue level and tissues suffer from hypoxia, which is the most potent stimulus for erythropoietin secretion. Since erythropoietin is increased there is increase in hemoglobin content in newborns.
5. What is physiological jaundice?
 - Bilirubin discoloration of skin and mucous membrane seen in newborn is called physiological jaundice. It disappears in 2 weeks. After birth fetal hemoglobin is replaced by adult hemoglobin. There is increased destruction of red blood cells after birth in order to bring the HbC count to normal level. This leads to increase in the level of bilirubin in blood leading to jaundice.
6. What is the principle behind Sahlb's method of hemoglobin estimation?
 - Refer page 28.
7. How is the standard color in Sahlb's method obtained?
 - The standard color is prepared by treating 20 mm³ of blood containing 14.5 g of hemoglobin per 100 mL with N/10 HCl and diluting it 100 times.
8. What is the reason for waiting for 10 min after adding blood into the diluting tube?
 - This much time is needed for the complete hemolysis of RBC and for the conversion of all the hemoglobin present to form acid hematin.
9. What are the functions of hemoglobin?
 - It helps in the transport of O₂ from lungs to tissues by forming oxyhemoglobin.
 - Helps in the transport of CO₂ from tissues to lungs by forming carbaminohemoglobin.
 - It acts as a buffer to maintain the normal pH of blood. Hemoglobin being a protein is responsible for 70% of the buffering capacity of whole blood.
 - The β-chain of hemoglobin has an additional site, called HbC binding site. The affinity of hemoglobin for NO is increased by O₂. So hemoglobin binds with NO in the lungs and releases it in the tissues where it promotes vasodilatation.
10. What are the various forms of hemoglobin present in blood?
 - Oxyhemoglobin (HbO₂)—Hb in combination with O₂.
 - Carbaminohemoglobin (HbCO)—Hb in combination with CO.
 - Carboxyhemoglobin (HbCO)—Hb in combination with CO.
 - Sulfhemoglobin (HbS)—Hb in combination with sulfur containing compounds.
 - Methemoglobin—when the ferrous form of iron is oxidized to the ferric form.
 - Adult hemoglobin are HbA (α₂β₂) and HbA₂ (α₂β₂).
 - Fetal hemoglobin are HbF (α₂γ₂).
11. What is anemia? How is it graded?
 - Reduction in the hemoglobin content of blood or RBC count or both below the normal range for that age and sex is called anemia. When the HbC count is less than 4 millimoles of blood or hemoglobin content less than 12 g/dL, in an adult the condition is anemia. Severity of anemia can be graded depending on the hemoglobin content as follows:
 - **Mild anemia** when the hemoglobin content is 10–11 g/L.
 - **Moderate anemia** when the hemoglobin content is 7–9 g/L.
 - **Severe anemia** when the hemoglobin content falls below 6 g/L.

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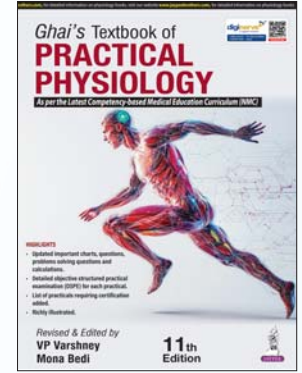
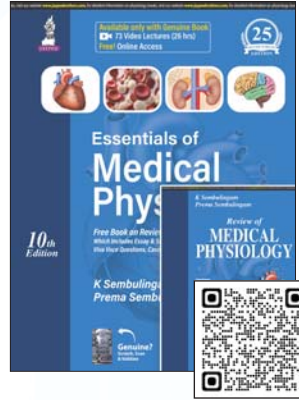
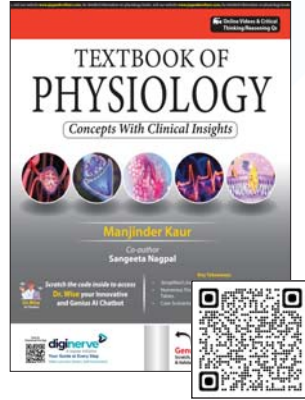
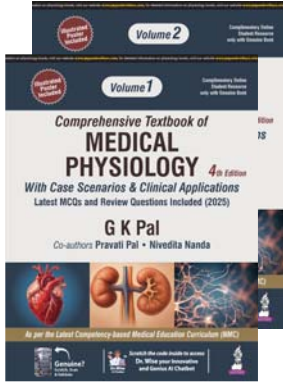
of jaundice. It disappears in 2 weeks after birth in order to bring the HbC count to normal level.

100 mL with N/10 HCl and diluting it 100 times.

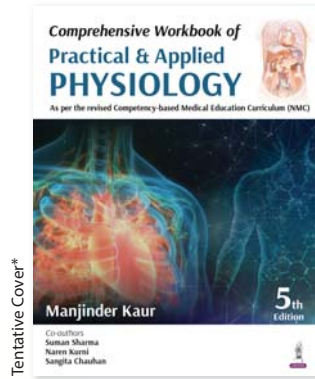
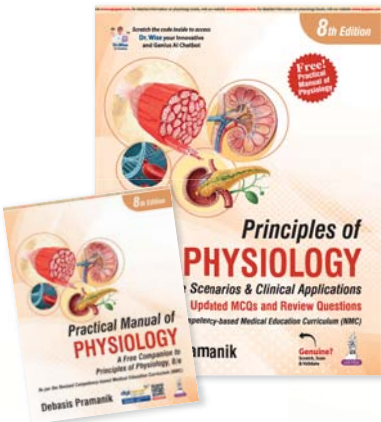
hemoglobin present to form acid hematin.

responsible for 70% of the buffering capacity of hemoglobin for NO is increased by O₂. So vasodilatation.

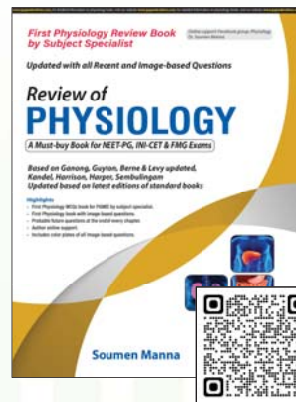
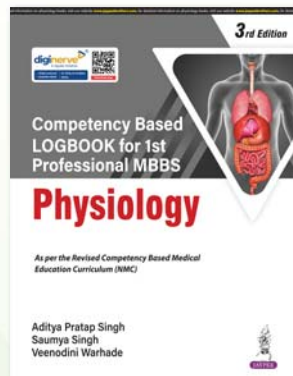
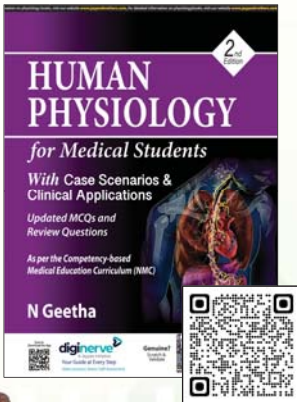
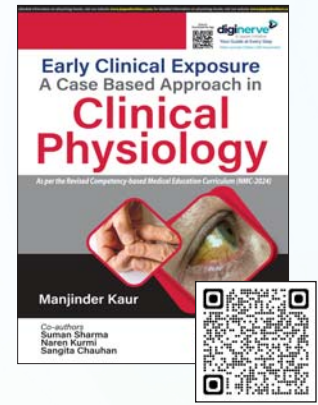
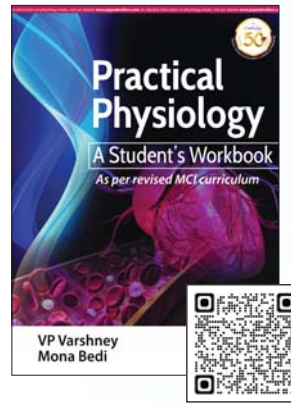
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